

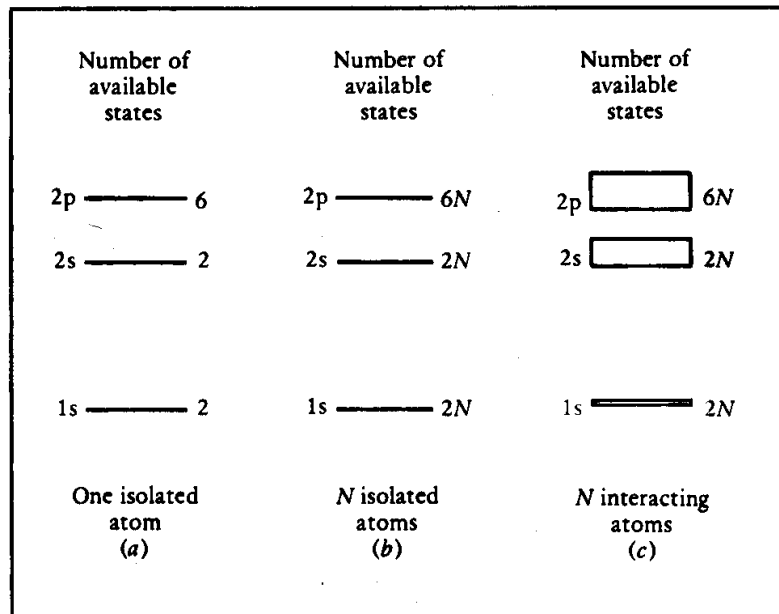
Maximum Theoretical Efficiency of PV Cells

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UC San Diego MAE 119

Lecture Notes

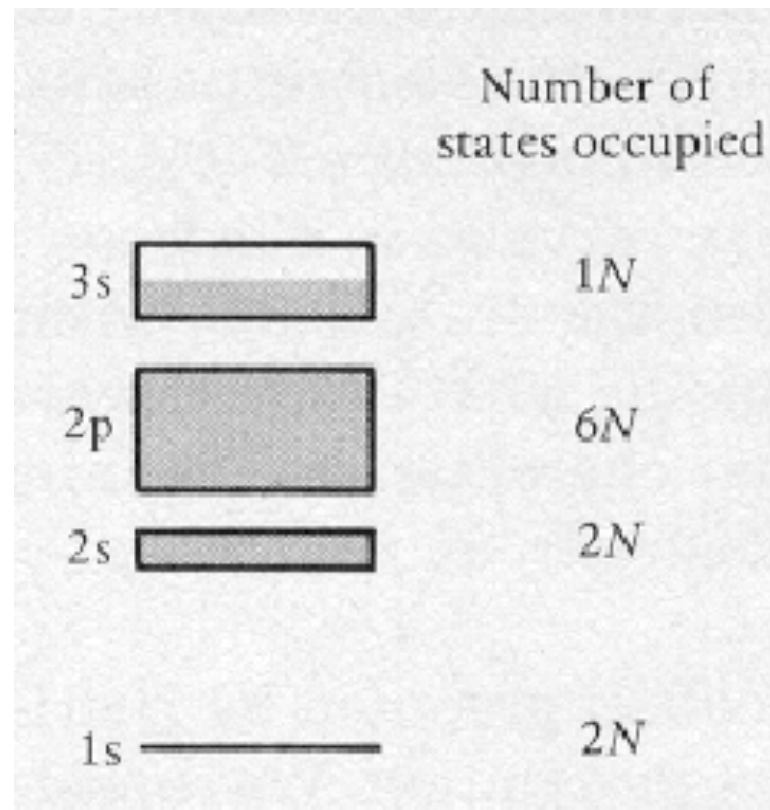
Band Theory of Solids: From Single Atoms...to Solid Crystals



- Isolated Li atom (conducting metal)
 - Has well-defined, isolated allowable electron energy levels
- N isolated atoms
 - $N \times$ isolated atom levels
- Strongly interacting Li atoms
 - Interaction shifts (or splits) individual **energy bands** into isolated regions separated by **forbidden bands**

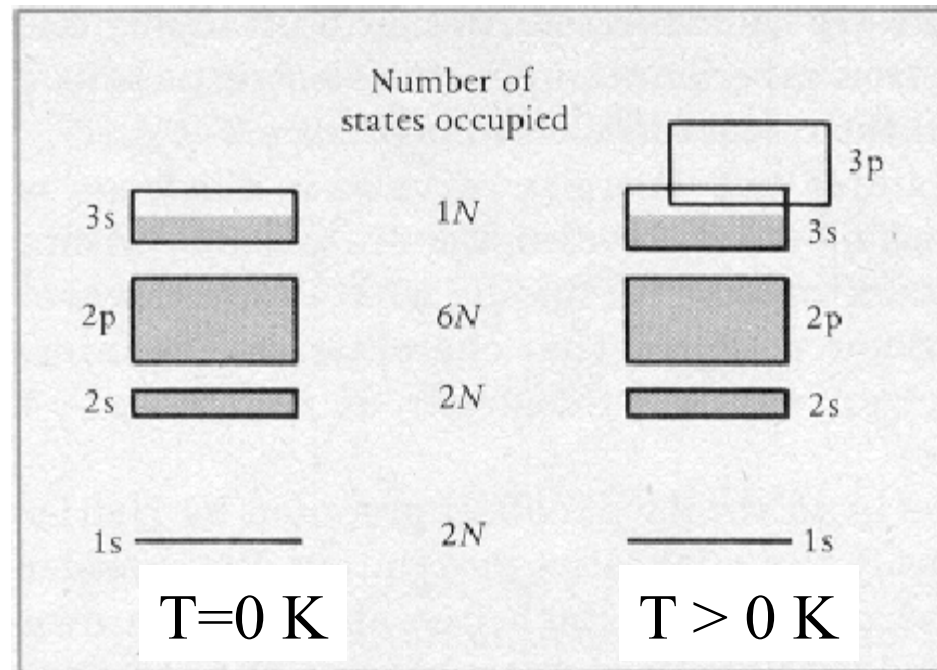
Band Theory of Solids: Conductors

- Next, consider N interacting sodium atoms at 0 deg K
 - Electrons in config $1s^2 2s^2 2p^6 3s^1$
 - Shells filled to 3s, which has 1 electron

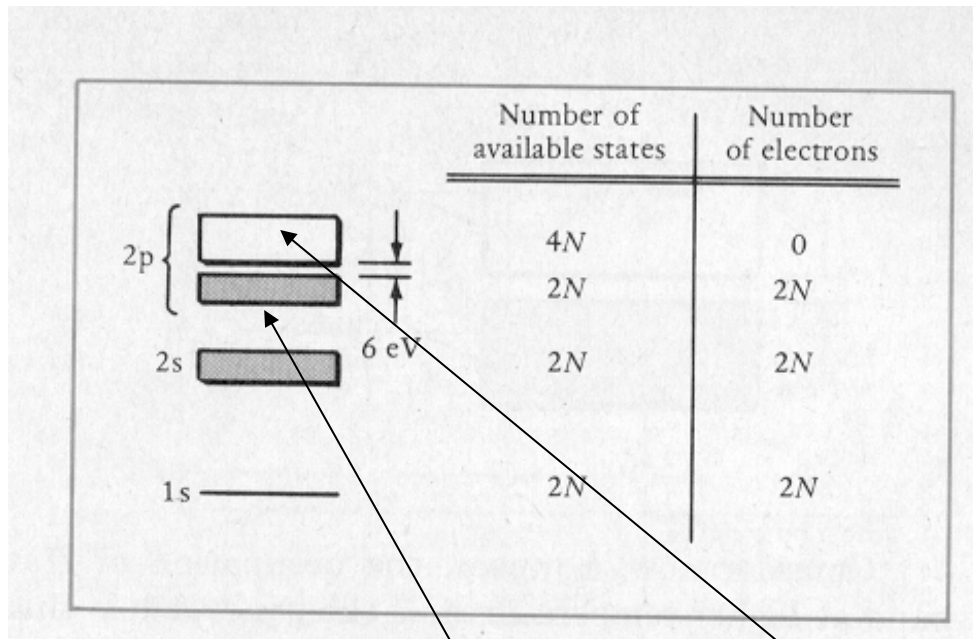


Band Theory of Solids: Conductors

- Next, consider N interacting sodium atoms w/ $T > 0$
 - Electrons in config $1s^2 2s^2 2p^6 3s^1$
 - Shells filled to $3s$, which has 1 electron
 - This Valence electron is weakly bound \Rightarrow if T High enough can move to mobile state \rightarrow conductor!



Band Theory of Solids: Insulators

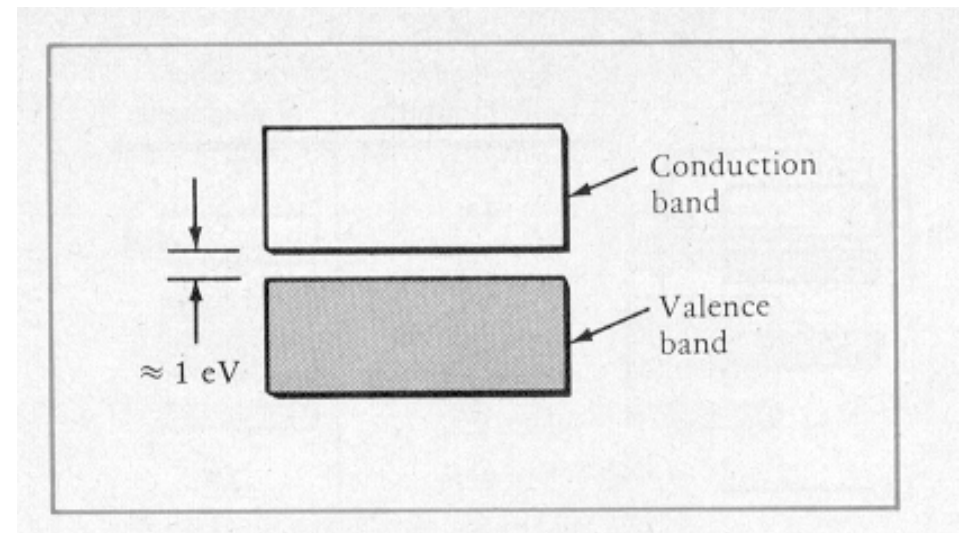


Lower levels: VALENCE BAND
Upper levels: CONDUCTION BAND

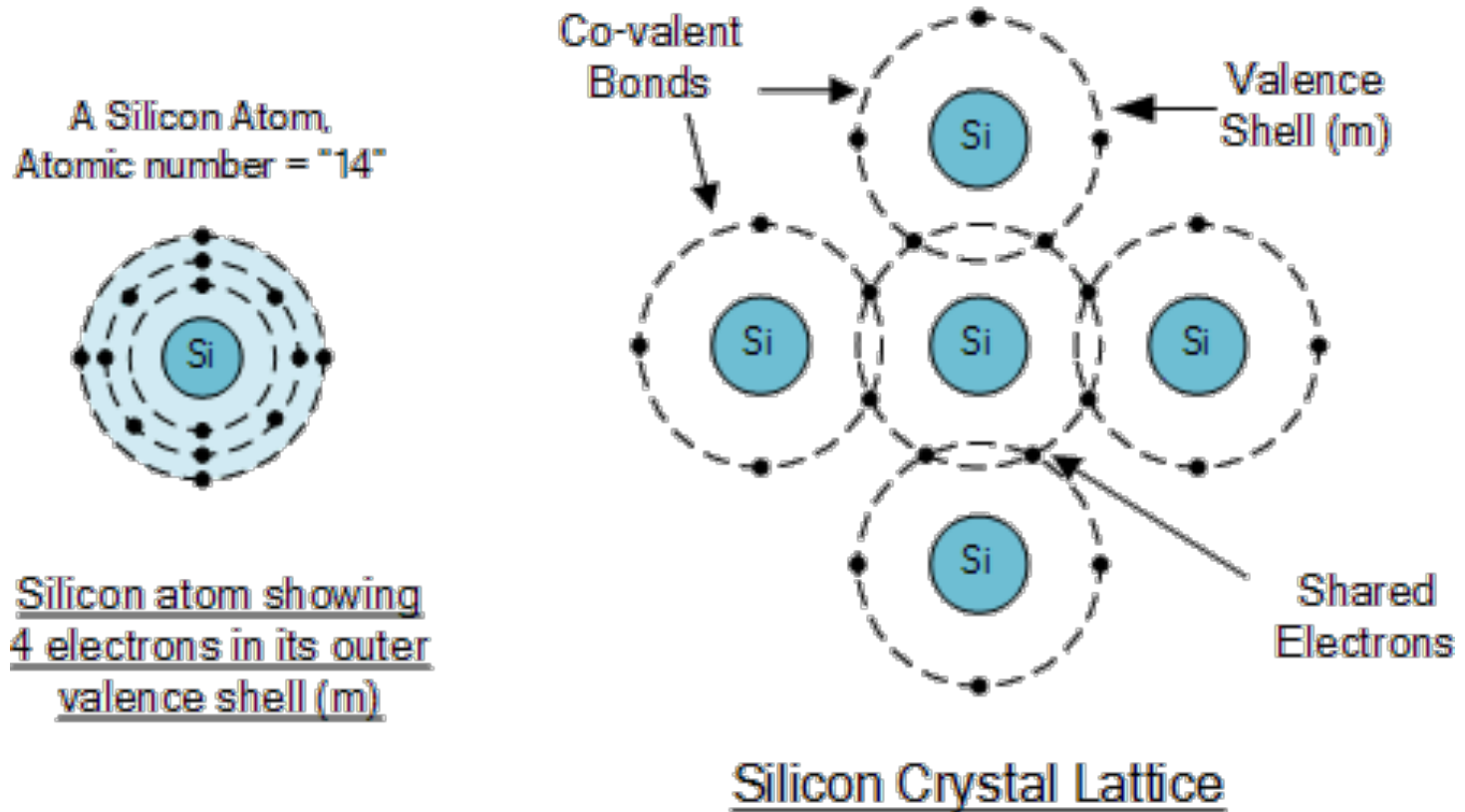
- Carbon in Diamond Form
 - Electrons in $1s^2 2s^2 2p^2$ State
 - 2p band has 2N electrons, but 6N states
 - BUT... crystal structure splits 2p into two distinct bands
 - **BAND GAP is ~ 6 eV**
 \gg Temperature
($\sim 0.02-0.1$ eV)
Thus...Diamond is
An Insulator

Band Theory of Solids: Semiconductors

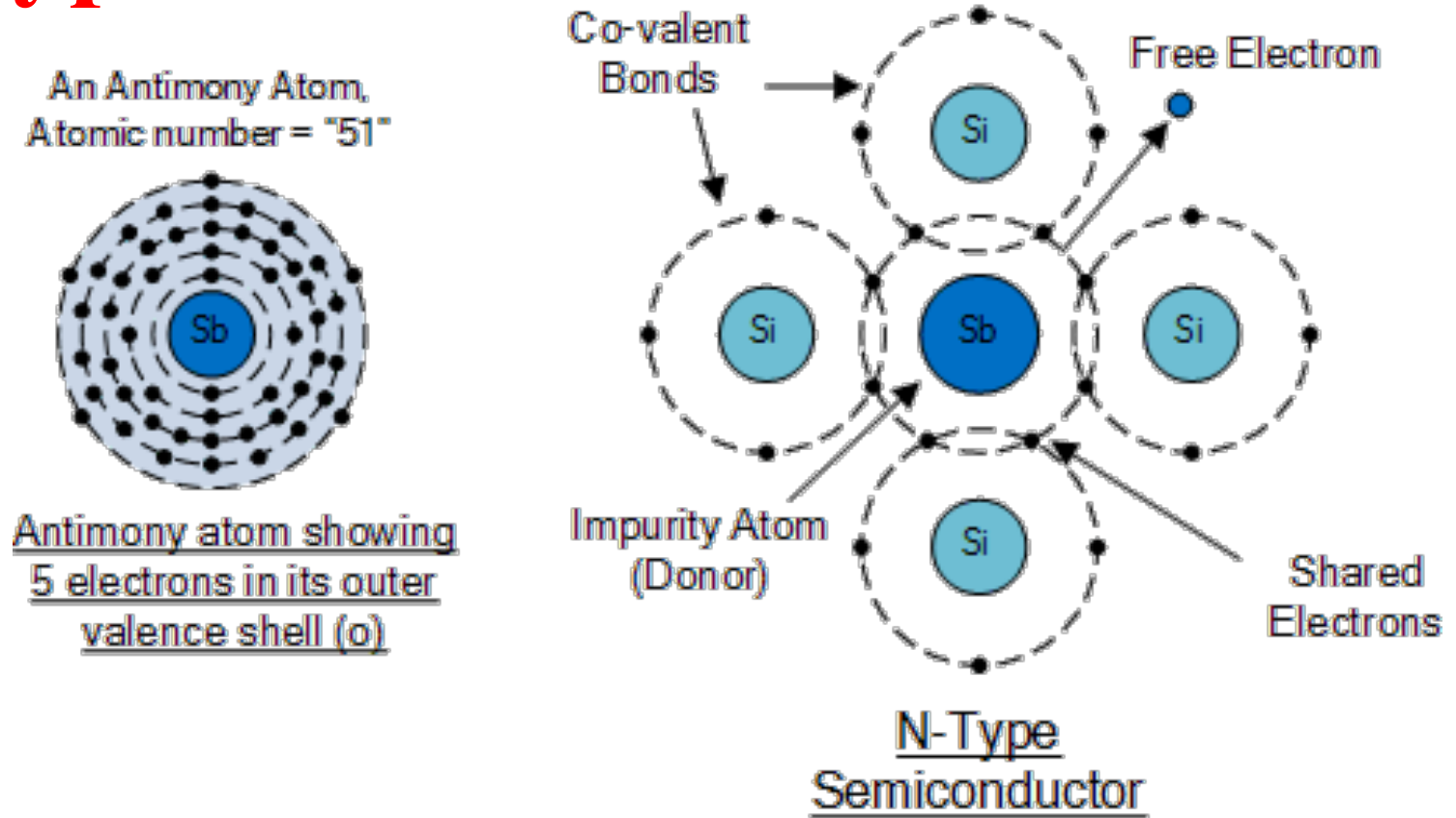
- *Some crystalline materials have smaller band-gap energy*
- At low temperatures behave like insulators
 - $E_{bg} \sim 1\text{eV} \gg \text{Temperature}$
- With an electric field
 - Electrons gain energy
 - Can move into upper (conduction) band



Si as a Semiconductor Material

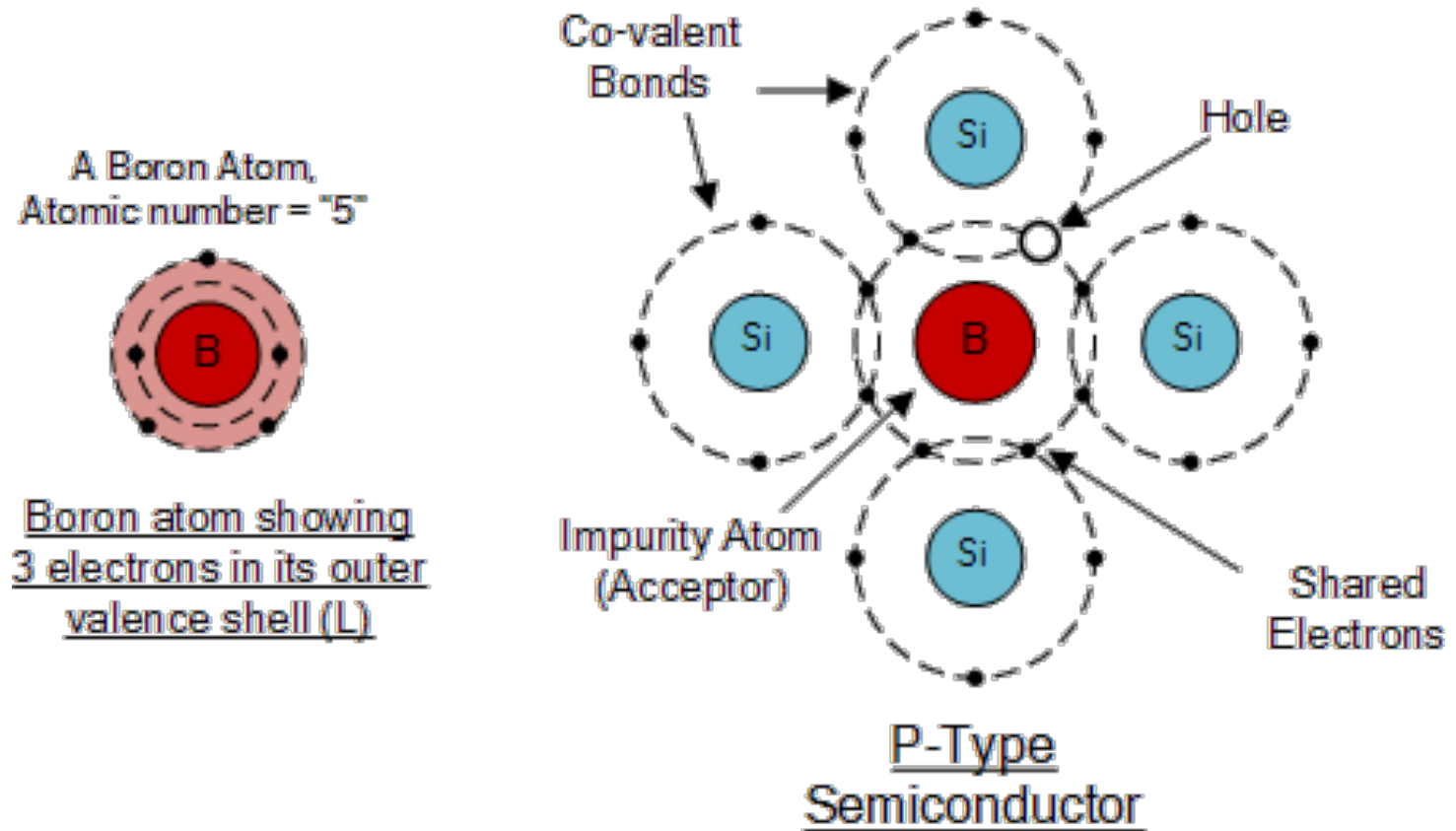


N-type Semiconductor Materials



N-type Si has an extra electron for each dopant atom,
This electron is mobile

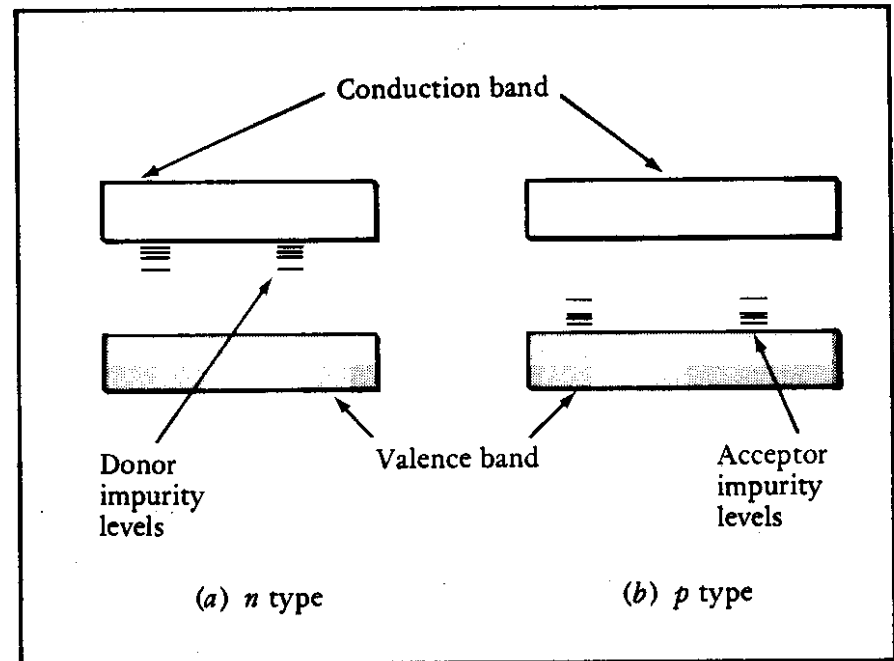
P-type Semiconductor Materials



P-type Si has a "hole" (i.e a missing electron) that acts like
A mobile positive charge

Dopants create allowed energy states between the pure material valence and conduction bands

- Pure semiconductor matl's conduction and valence bands separated by E_{gap}
- In pure materials this gap has no allowed states -> no particles in these energy ranges
- IF ADD donor or acceptor impurities then this creates allowed states between the pure-material conduction & valence bands



Physics of maximum theoretical efficiency

- Key Concepts
 - Photon Energy Spectrum
 - Charge Carrier Generation Via Photon Absorption
 - Photon flux & relation to energy spectrum
 - Estimating maximum possible efficiency
 - What does a real cell look like?

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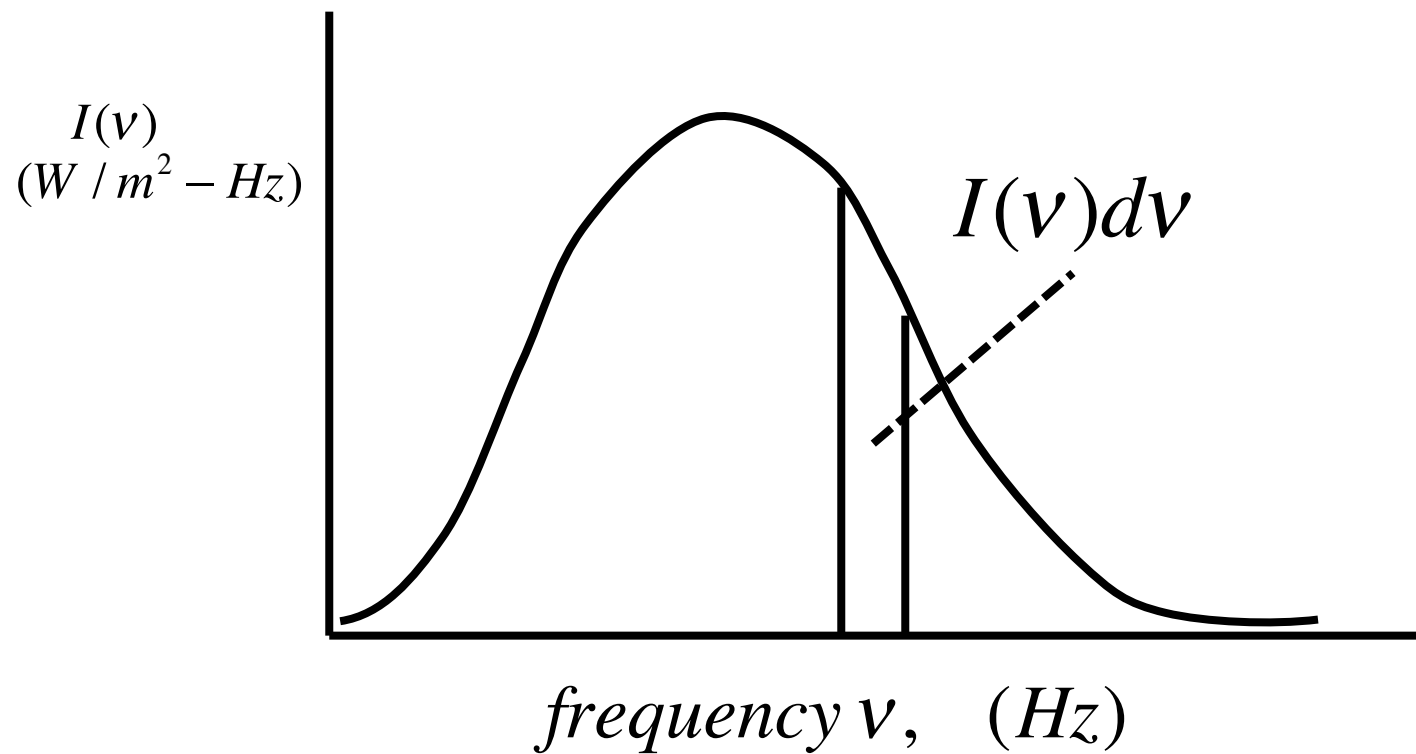
Recall blackbody spectrum:

Blackbody spectrum:

$$I(\nu) = \frac{h\nu^3}{c^2} \frac{1}{\exp(h\nu / kT) - 1}$$

Total intensity:

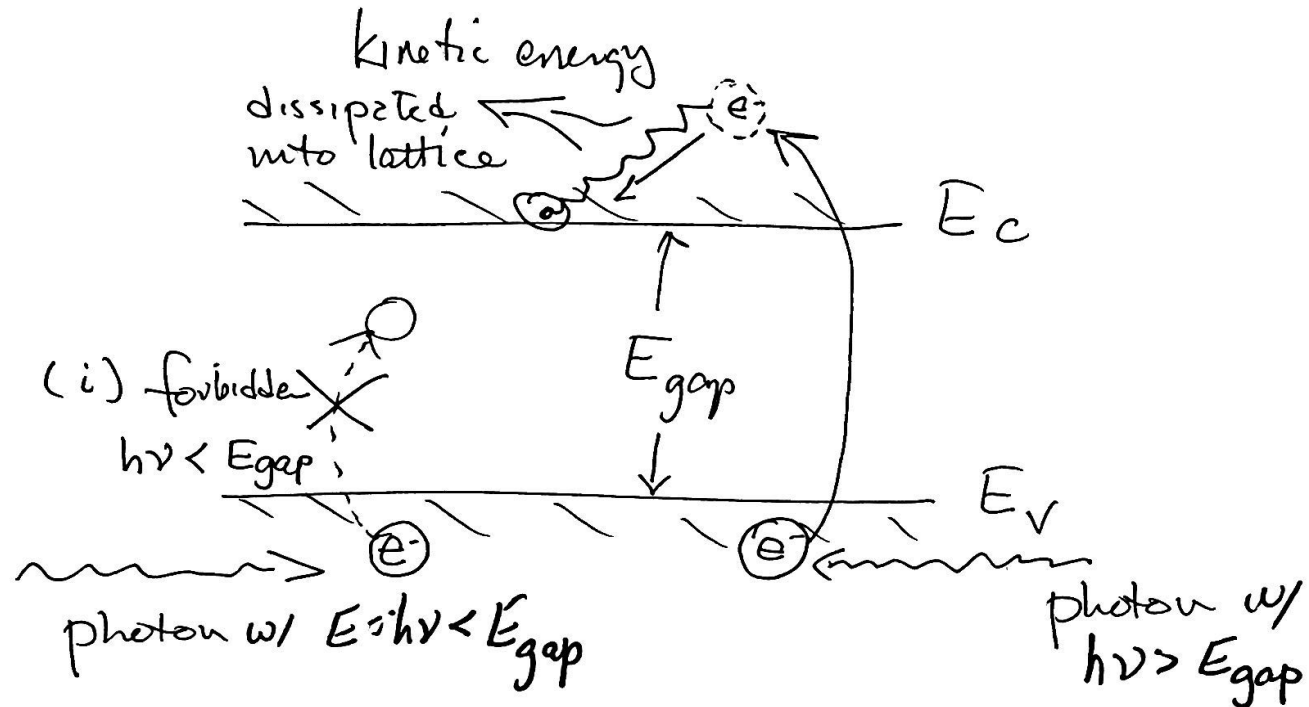
$$I_{tot} = \int_0^{\infty} I(\nu) d\nu$$



Physics of maximum theoretical efficiency

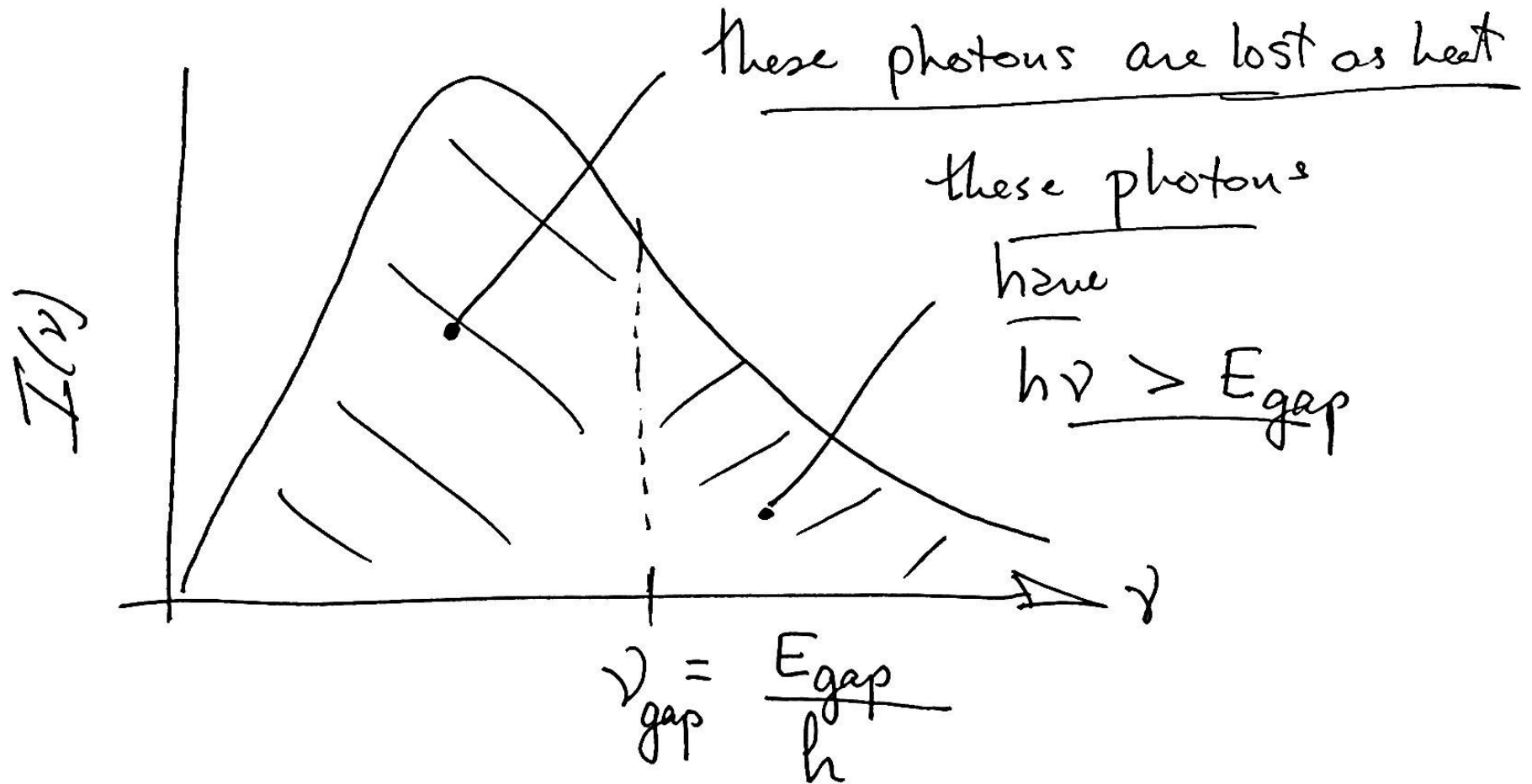
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Light absorption can (sometimes) create mobile e-h pairs:



(i) if $E_{photon} = h\nu < E_{gap} \rightarrow$ photon cannot dislodge e^- from V -band

What portion of spectrum has photons w/ enough energy?



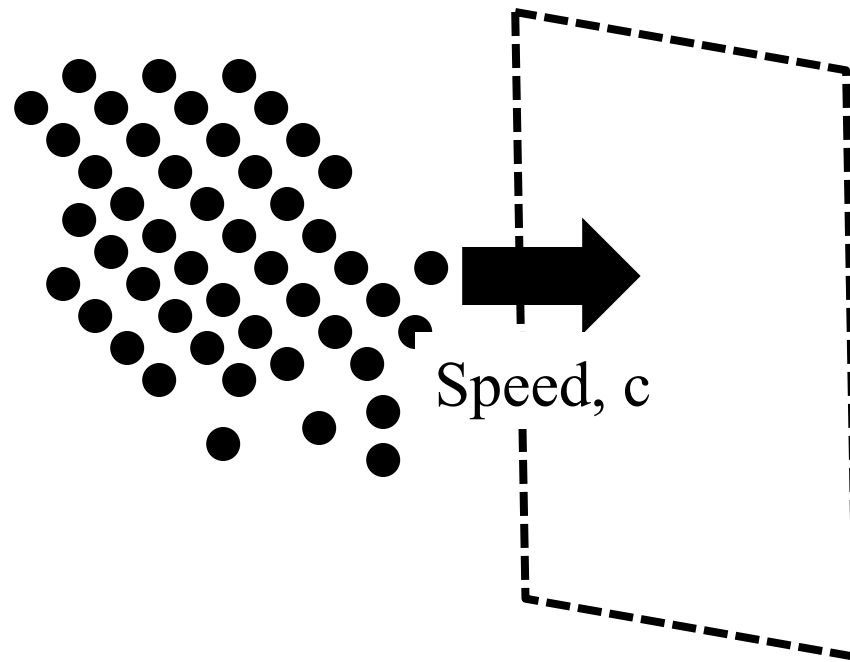
Physics of maximum theoretical efficiency

- Key Concepts
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Relation of photon flux to Intensity:

Collection of n photons/unit volume

With frequency in range $(\nu, \nu + d\nu)$

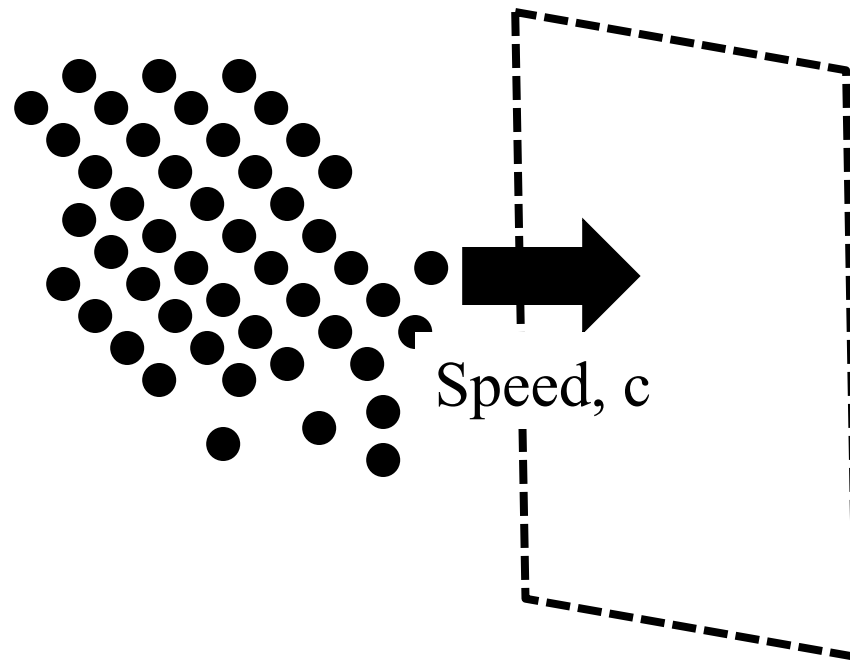


Q: How many photons pass thru the surface per unit area/unit time?

Relation of photon flux to Intensity:

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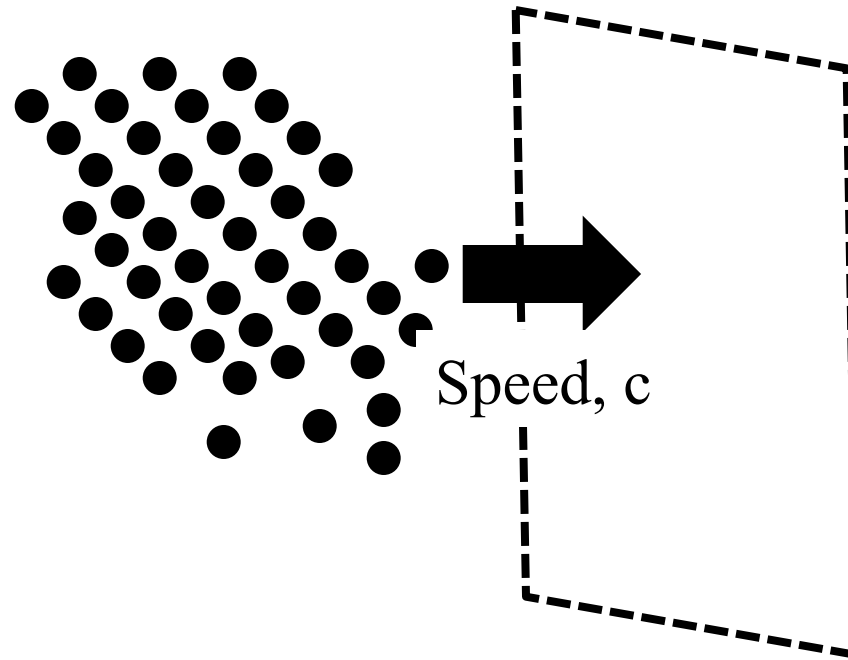
Q: How many photons pass thru the surface per unit area/unit time?

A: This is the photon flux, $\phi(\nu) = n(\nu)c$

Relation of photon flux to Intensity:

Collection of n photons/unit volume

With frequency in range $(\nu, \nu + d\nu)$

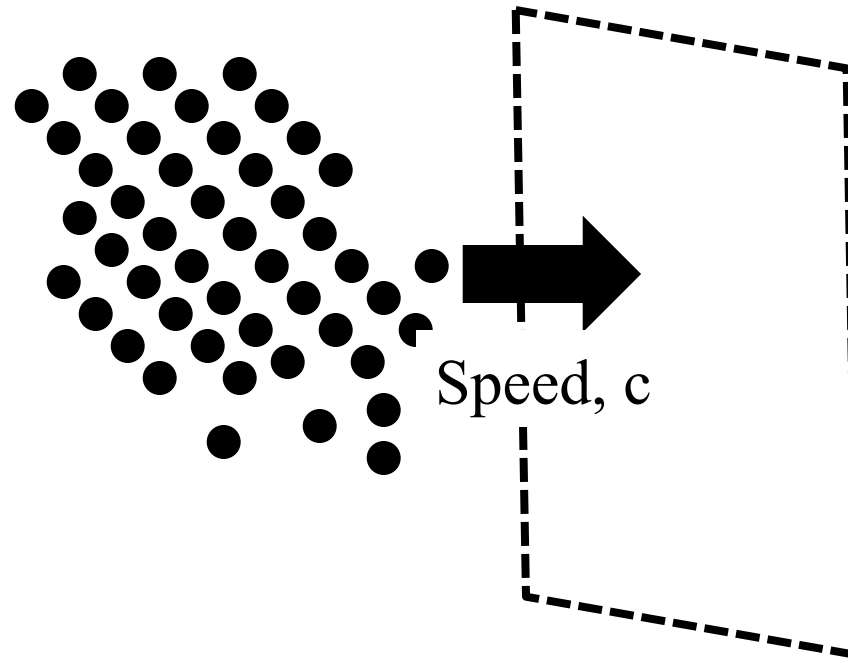


Q: If each photon has energy $E=h\nu$, how much energy passes thru surface per unit area and per unit time ?

Relation of photon flux to Intensity:

Collection of n photons/unit volume

With frequency in range $(\nu, \nu + d\nu)$



Q: If each photon has energy $E=h\nu$, how much energy passes thru surface per unit area and per unit time ?

A: Energy per unit area/unit time is INTENSITY,

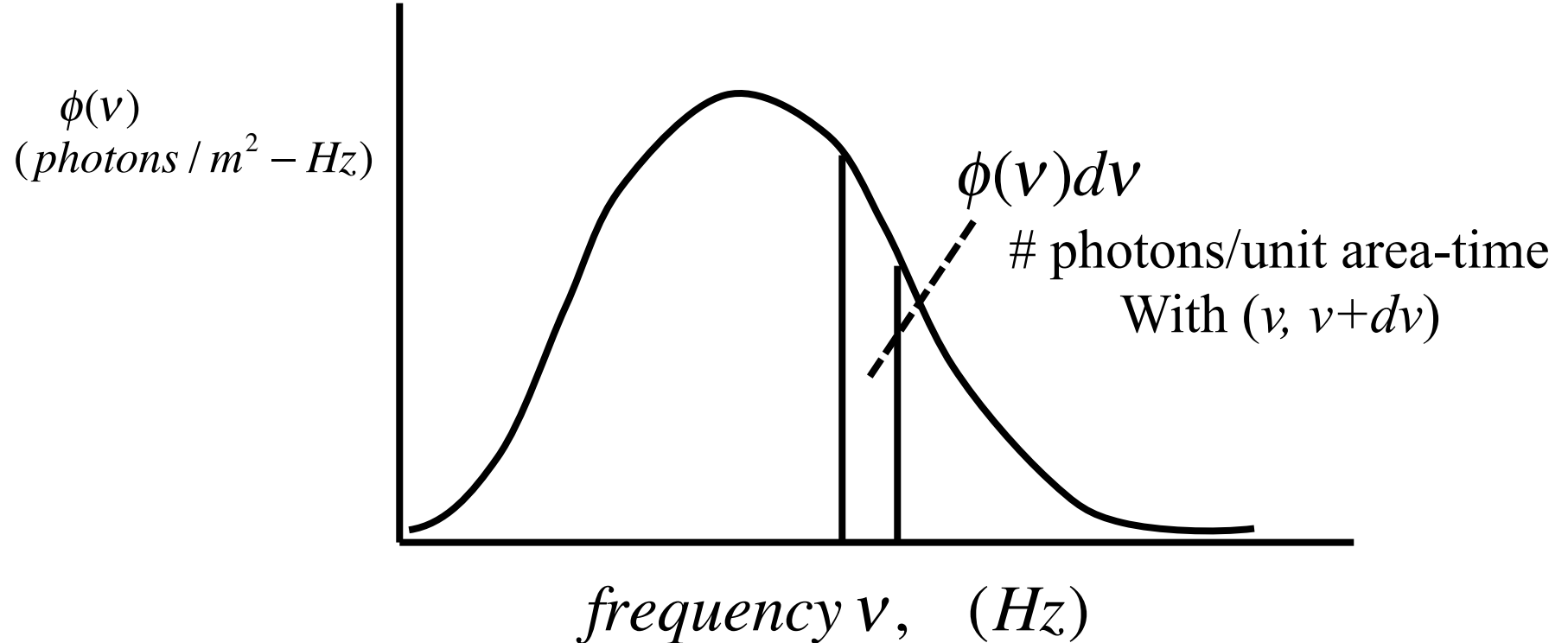
$$I(\nu) = h\nu\phi(\nu)$$

Photon flux of blackbody spectrum:

Blackbody spectrum:

$$\phi(\nu) = \frac{\nu^2}{c^2} \frac{1}{\exp(h\nu / kT) - 1}$$

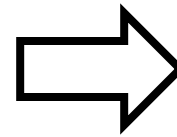
Total flux (photons/unit area-time):



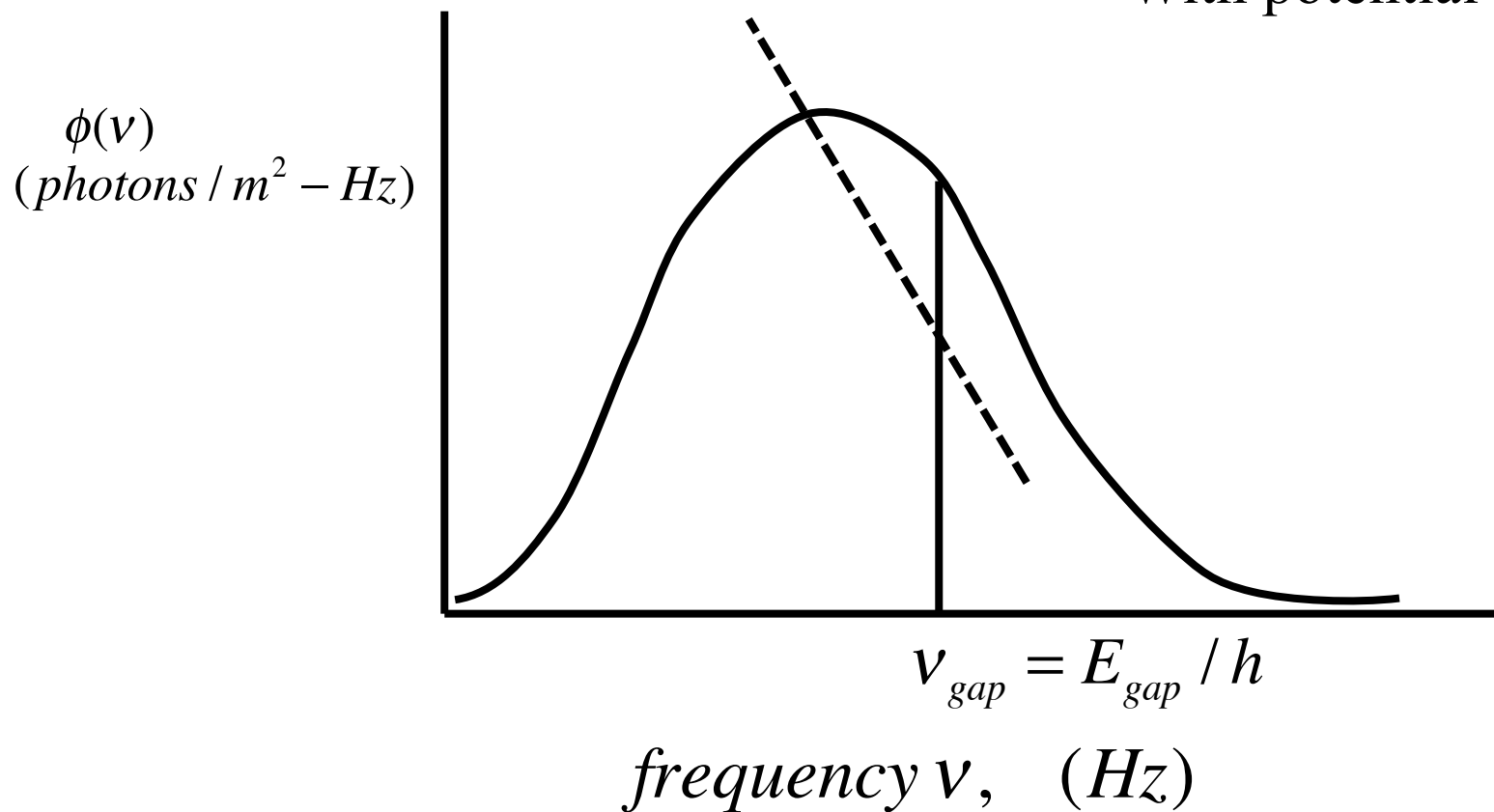
Photon flux of blackbody spectrum:

$$\phi_{gap} = \int_{\nu_{gap}}^{\infty} \phi(\nu) d\nu$$

Photon flux with $E > E_{gap}$



Each of these
Photons produce
Charge carrier pair
With potential energy E_{gap}



Can now estimate max efficiency:

Power produced by e-h pair creation:

$$P_{\max} = \phi_{gap} E_{gap} \quad \text{Where} \quad \phi_{gap} = \int_{v_{gap}}^{\infty} \phi(v) dv$$

Maximum incident power (per unit area):

$$I_{tot} = \int_0^{\infty} I(v) dv$$

Maximum possible efficiency is the ratio of these two:

$$\eta_{\max} = \frac{P_{\max}}{I_{tot}}$$

Max. PV Cell Efficiency:

Can recast as an integral:

$$\eta_{\max} = \frac{15}{\pi^4} \xi_0 \int_{\xi_0}^{\infty} \frac{x^2}{e^x - 1} dx$$

where $\xi_0 = \frac{qE_{\text{gap}}}{k_B T_{\text{bb}}}$

For Si with $E_{\text{gap}} \sim 1.1$ eV and $T_{\text{bb}} \sim 6000$ K

$$\eta_{\max} \sim 0.44$$

A Solar PV Cell is just a p-n junction (“diode”) illuminated by light....

