

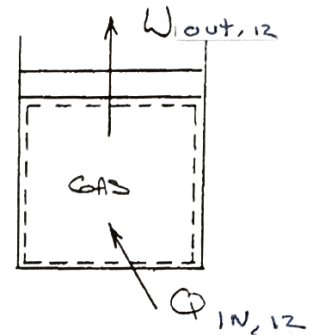
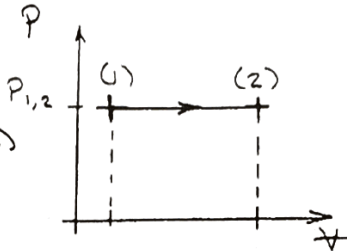
1.1

Problem Description:

A frictionless piston is raised slowly by heating the gas contained in the cylinder.

GIVEN:

$$\begin{aligned}
 p_1 &= 0.2 \text{ MPa} \\
 p_2 &= p_1 \text{ (constant pressure)} \\
 v_1 &= 1.0 \text{ m}^3 \\
 v_2 &= 2.0 \text{ m}^3 \\
 Q_{1,2} &= 2000 \text{ kJ} = 2 \text{ MJ}
 \end{aligned}$$



DETERMINE: CHANGE IN INTERNAL ENERGY, ΔU_{12}

ENGINEERING MODEL:

- ① QUASIEQUILIBRIUM PROCESS
- ② SYSTEM IS THE GAS ONLY.
- ③ NEGUGIBLE KINETIC AND POTENTIAL ENERGY EFFECTS.

BASIC EQUATIONS:

$$1^{\text{st}} \text{ LAW FOR CLOSED SYSTEM: } \Delta KE + \Delta PE + \Delta U = Q_{IN} + W_{IN} - Q_{OUT} - W_{OUT} \quad (1)$$

$$\text{Work: } W_{12} = \int_1^2 p \, dv \quad (2)$$

STEPS:

Apply 1st law to system:

$$Q_{IN,12} = W_{out,12} + \cancel{\Delta KE} + \cancel{\Delta PE} + \Delta U_{12}$$

$$\text{SOLVING FOR } \Delta U_{12}: \quad \Delta U_{12} = Q_{IN,12} - W_{out,12} \quad (3)$$

To determine $W_{out,12}$, use Eq. (2):

$$W_{out,12} = |W_{12}| = \left| \int_1^2 p \, dv \right|$$

$$W_{out,12} = |p(v_2 - v_1)| \quad (4)$$

SUBSTITUTING (4) INTO (3):

$$\Rightarrow \Delta U_{12} = U_2 - U_1 = Q_{IN,12} - |p(v_2 - v_1)| \quad (5)$$

Numerical Substitution:

$$\text{Eq. (4)} \rightarrow W_{out,12} = 0.2 \text{ MPa} (2. - 1.) \text{ m}^3 \left(\frac{1 \text{ N/m}^2}{1 \text{ Pa}} \right) \left(\frac{1 \text{ J}}{1 \text{ N}\cdot\text{m}} \right) = 0.2 \text{ MJ}$$

$$\text{Eq. (5)} \rightarrow \Delta U_{12} = 2 - 0.2 = \boxed{1.8 \text{ MJ}}$$