

Reference Sheet IV

Some of the important equations in this chapter are:

Conservation of mass
$$\frac{\partial}{\partial t} \int_{cv} \rho dV + \int_{cs} \rho \mathbf{V} \cdot \hat{\mathbf{n}} dA = 0 \quad (5.5)$$

Mass flowrate
$$\dot{m} = \rho Q = \rho AV \quad (5.6)$$

First law of thermodynamics (Conservation of energy)
$$\frac{\partial}{\partial t} \int_{cv} e \rho dV + \int_{cs} \left(\check{u} + \frac{p}{\rho} + \frac{V^2}{2} + gz \right) \rho \mathbf{V} \cdot \hat{\mathbf{n}} dA = \dot{Q}_{net, in} + \dot{W}_{shaft, net, in} \quad (5.64)$$

Some of the important equations in this chapter are:

Ideal gas equation of state
$$\rho = \frac{p}{RT} \quad (11.1)$$

Internal energy change
$$\check{u}_2 - \check{u}_1 = c_v(T_2 - T_1) \quad (11.5)$$

Enthalpy
$$\check{h} = \check{u} + \frac{p}{\rho} \quad (11.6)$$

Enthalpy change
$$\check{h}_2 - \check{h}_1 = c_p(T_2 - T_1) \quad (11.9)$$

Specific heat difference
$$c_p - c_v = R \quad (11.12)$$

Specific heat ratio
$$k = \frac{c_p}{c_v} \quad (11.13)$$

Specific heat at constant pressure
$$c_p = \frac{Rk}{k-1} \quad (11.14)$$

Specific heat at constant volume
$$c_v = \frac{R}{k-1} \quad (11.15)$$

First Tds equation
$$T ds = d\check{u} + pd \left(\frac{1}{\rho} \right) \quad (11.16)$$

Second Tds equation
$$T ds = d\check{h} - \left(\frac{1}{\rho} \right) dp \quad (11.18)$$

Entropy change
$$s_2 - s_1 = c_v \ln \frac{T_2}{T_1} + R \ln \frac{\rho_1}{\rho_2} \quad (11.21)$$

Entropy change
$$s_2 - s_1 = c_p \ln \frac{T_2}{T_1} - R \ln \frac{p_2}{p_1} \quad (11.22)$$

Isentropic flow
$$\frac{p}{\rho^k} = \text{constant} \quad (11.25)$$

Speed of sound
$$c = \sqrt{\left(\frac{\partial p}{\partial \rho} \right)_s} \quad (11.34)$$

Speed of sound in gas
$$c = \sqrt{RTk} \quad (11.36)$$

Speed of sound in liquid
$$c = \sqrt{\frac{E_v}{\rho}} \quad (11.38)$$

Mach cone angle
$$\sin \alpha = \frac{c}{V} = \frac{1}{Ma} \quad (11.39)$$

Mach number
$$Ma = \frac{V}{c} \quad (11.46)$$

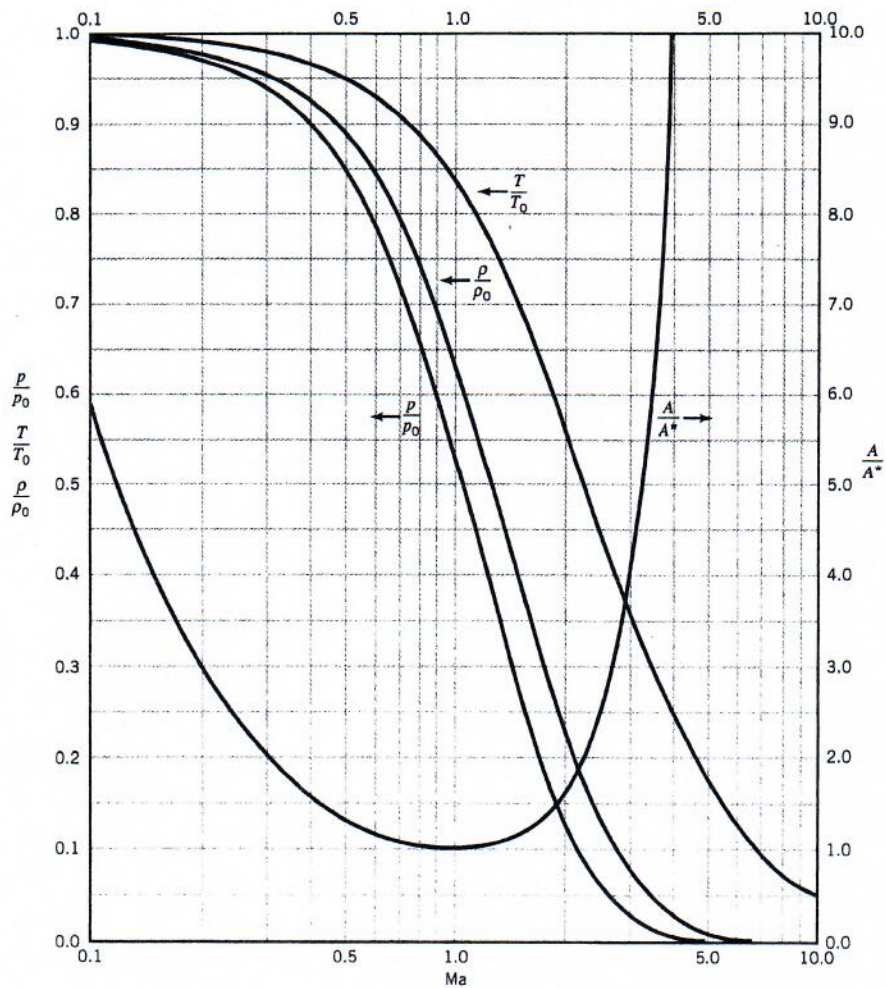
Isentropic Flow Relations (Ideal Gas with $k = 1.4$)

Isentropic flow	$\frac{T}{T_0} = \frac{1}{1 + [(k - 1)/2]Ma^2}$	(11.56)
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Isentropic flow	$\frac{p}{p_0} = \left\{ \frac{1}{1 + [(k - 1)/2]Ma^2} \right\}^{k/(k-1)}$	(11.59)
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Isentropic flow	$\frac{\rho}{\rho_0} = \left\{ \frac{1}{1 + [(k - 1)/2]Ma^2} \right\}^{1/(k-1)}$	(11.60)
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Isentropic flow	$\frac{A}{A^*} = \frac{1}{Ma} \left\{ \frac{1 + [(k - 1)/2]Ma^2}{1 + [(k - 1)/2]} \right\}^{(k+1)/[2(k-1)]}$	(11.71)
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■ Figure D.1 Isentropic flow of an ideal gas with $k = 1.4$. (Graph provided by Dr. Bruce A. Reichert.)

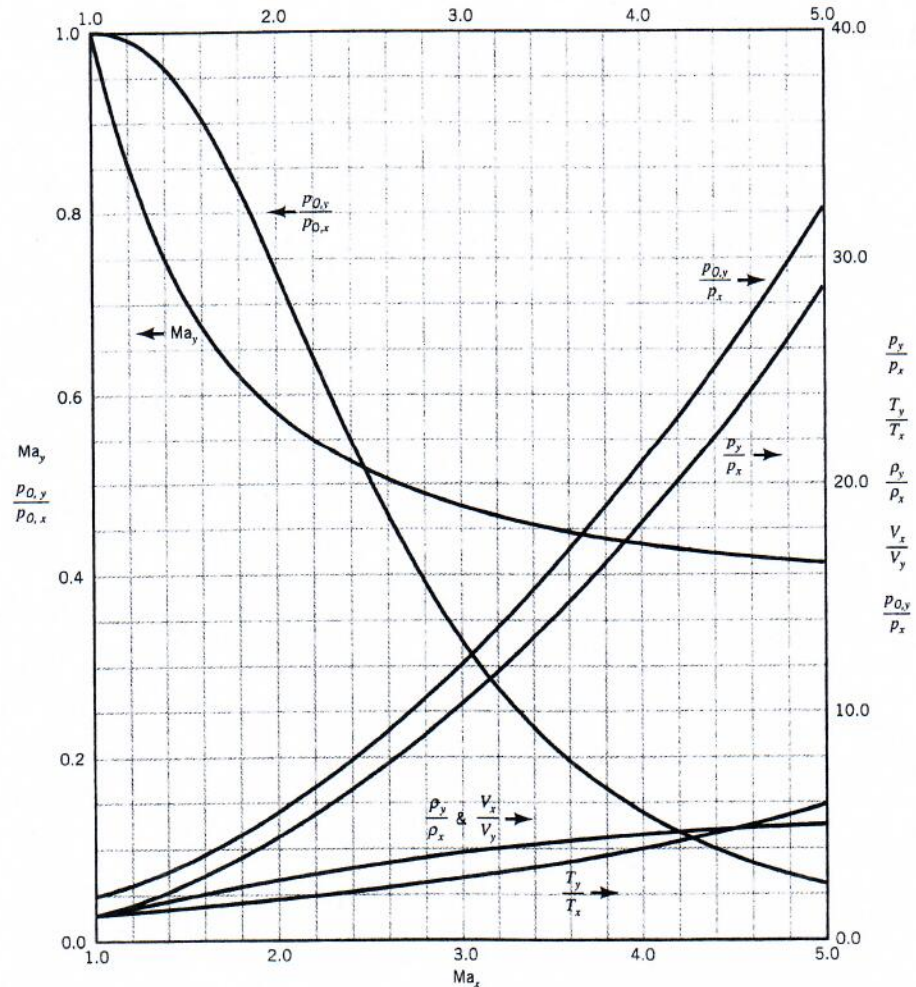
■ Table D.1

Isentropic Flow Functions for an Ideal Gas with $k = 1.4$

Ma	T/T_0	p/p_0	ρ/ρ_0	A/A^*	Ma	T/T_0	p/p_0	ρ/ρ_0	A/A^*
0.0000	1.0000	1.0000	1.0000	∞	2.0500	0.5433	0.1182	0.2176	1.7600
0.0500	0.9995	0.9983	0.9988	11.5914	2.1000	0.5313	0.1094	0.2058	1.8369
0.1000	0.9980	0.9930	0.9950	5.8218	2.1500	0.5196	0.1011	0.1946	1.9185
0.1500	0.9955	0.9844	0.9888	3.9103	2.2000	0.5081	0.0935	0.1841	2.0050
0.2000	0.9921	0.9725	0.9803	2.9635	2.2500	0.4969	0.0865	0.1740	2.0964
0.2500	0.9877	0.9575	0.9694	2.4027	2.3000	0.4859	0.0800	0.1646	2.1931
0.3000	0.9823	0.9395	0.9564	2.0351	2.3500	0.4752	0.0740	0.1556	2.2953
0.3500	0.9761	0.9188	0.9413	1.7780	2.4000	0.4647	0.0684	0.1472	2.4031
0.4000	0.9690	0.8956	0.9243	1.5901	2.4500	0.4544	0.0633	0.1392	2.5168
0.4500	0.9611	0.8703	0.9055	1.4487	2.5000	0.4444	0.0585	0.1317	2.6367
0.5000	0.9524	0.8430	0.8852	1.3398	2.5500	0.4347	0.0542	0.1246	2.7630
0.5500	0.9430	0.8142	0.8634	1.2549	2.6000	0.4252	0.0501	0.1179	2.8960
0.6000	0.9328	0.7840	0.8405	1.1882	2.6500	0.4159	0.0464	0.1115	3.0359
0.6500	0.9221	0.7528	0.8164	1.1356	2.7000	0.4068	0.0430	0.1056	3.1830
0.7000	0.9107	0.7209	0.7916	1.0944	2.7500	0.3980	0.0398	0.0999	3.3377
0.7500	0.8989	0.6886	0.7660	1.0624	2.8000	0.3894	0.0368	0.0946	3.5001
0.8000	0.8865	0.6560	0.7400	1.0382	2.8500	0.3810	0.0341	0.0896	3.6707
0.8500	0.8737	0.6235	0.7136	1.0207	2.9000	0.3729	0.0317	0.0849	3.8498
0.9000	0.8606	0.5913	0.6870	1.0089	2.9500	0.3649	0.0293	0.0804	4.0376
0.9500	0.8471	0.5595	0.6604	1.0021	3.0000	0.3571	0.0272	0.0762	4.2346
1.0000	0.8333	0.5283	0.6339	1.0000	3.1000	0.3422	0.0234	0.0685	4.6573
1.0500	0.8193	0.4979	0.6077	1.0020	3.2000	0.3281	0.0202	0.0617	5.1210
1.1000	0.8052	0.4684	0.5817	1.0079	3.3000	0.3147	0.0175	0.0555	5.6286
1.1500	0.7908	0.4398	0.5562	1.0175	3.4000	0.3019	0.0151	0.0501	6.1837
1.2000	0.7764	0.4124	0.5311	1.0304	3.5000	0.2899	0.0131	0.0452	6.7896
1.2500	0.7619	0.3861	0.5067	1.0468	3.6000	0.2784	0.0114	0.0409	7.4501
1.3000	0.7474	0.3609	0.4829	1.0663	3.7000	0.2675	0.0099	0.0370	8.1691
1.3500	0.7329	0.3370	0.4598	1.0890	3.8000	0.2572	0.0086	0.0335	8.9506
1.4000	0.7184	0.3142	0.4374	1.1149	3.9000	0.2474	0.0075	0.0304	9.7990
1.4500	0.7040	0.2927	0.4158	1.1440	4.0000	0.2381	0.0066	0.0277	10.7188
1.5000	0.6897	0.2724	0.3950	1.1762	4.5000	0.1980	0.0035	0.0174	16.5622
1.5500	0.6754	0.2533	0.3750	1.2116	5.0000	0.1667	0.0019	0.0113	25.0000
1.6000	0.6614	0.2353	0.3557	1.2502	5.5000	0.1418	0.0011	0.0076	36.8690
1.6500	0.6475	0.2184	0.3373	1.2922	6.0000	0.1220	0.0006	0.0052	53.1798
1.7000	0.6337	0.2026	0.3197	1.3376	6.5000	0.1058	0.0004	0.0036	75.1343
1.7500	0.6202	0.1878	0.3029	1.3865	7.0000	0.0926	0.0002	0.0026	104.1429
1.8000	0.6068	0.1740	0.2868	1.4390	7.5000	0.0816	0.0002	0.0019	141.8415
1.8500	0.5936	0.1612	0.2715	1.4952	8.0000	0.0725	0.0001	0.0014	190.1094
1.9000	0.5807	0.1492	0.2570	1.5553	8.5000	0.0647	0.0001	0.0011	251.0862
1.9500	0.5680	0.1381	0.2432	1.6193	9.0000	0.0581	0.0000	0.0008	327.1893
2.0000	0.5556	0.1278	0.2300	1.6875	10.0000	0.0476	0.0000	0.0005	535.9375

Normal Shock Relations (Ideal Gas with $k = 1.4$)

Normal shock	$\text{Ma}_y^2 = \frac{\text{Ma}_x^2 + [2/(k-1)]}{[2k/(k-1)]\text{Ma}_x^2 - 1} \quad (11.149)$
Normal shock	$\frac{p_y}{p_x} = \frac{2k}{k+1} \text{Ma}_x^2 - \frac{k-1}{k+1} \quad (11.150)$
Normal shock	$\frac{T_y}{T_x} = \frac{\{1 + [(k-1)/2]\text{Ma}_x^2\} \{[2k/(k-1)]\text{Ma}_x^2 - 1\}}{\{(k+1)^2/[2(k-1)]\}\text{Ma}_x^2} \quad (11.151)$
Normal shock	$\frac{\rho_y}{\rho_x} = \frac{V_x}{V_y} = \frac{(k+1)\text{Ma}_x^2}{(k-1)\text{Ma}_x^2 + 2} \quad (11.154)$
Normal shock	$\frac{p_{0,y}}{p_{0,x}} = \frac{\left(\frac{k+1}{2} \text{Ma}_x^2\right)^{k/(k-1)} \left(1 + \frac{k-1}{2} \text{Ma}_x^2\right)^{k/(1-k)}}{\left(\frac{2k}{k+1} \text{Ma}_x^2 - \frac{k-1}{k+1}\right)^{1/(k-1)}} \quad (11.156)$



■ **Figure D.4** Normal shock flow of an ideal gas with $k = 1.4$. (Graph provided by Dr. Bruce A. Reichert.)

Table D.S

Normal Shock Functions for an Ideal Gas with $k = 1.4$

Ma_x	Ma_y	p_y/p_x	T_y/T_x	$\rho_y/\rho_x = V_x/V_y$	p_{0y}/p_{0x}	Ma_x	Ma_y	p_y/p_x	T_y/T_x	$\rho_y/\rho_x = V_x/V_y$	p_{0y}/p_{0x}
1.00	1.0000	1.0000	1.0000	1.0000	1.0000	3.05	0.4723	10.6863	2.7383	3.9025	0.3145
1.05	0.9531	1.1196	1.0328	1.0840	0.9999	3.10	0.4695	11.0450	2.7986	3.9466	0.3012
1.10	0.9118	1.2450	1.0649	1.1691	0.9989	3.15	0.4669	11.4096	2.8598	3.9896	0.2885
1.15	0.8750	1.3763	1.0966	1.2550	0.9967	3.20	0.4643	11.7800	2.9220	4.0315	0.2762
1.20	0.8422	1.5133	1.1280	1.3416	0.9928	3.25	0.4619	12.1563	2.9851	4.0723	0.2645
1.25	0.8126	1.6563	1.1594	1.4286	0.9871	3.30	0.4596	12.5383	3.0492	4.1120	0.2533
1.30	0.7860	1.8050	1.1909	1.5157	0.9794	3.35	0.4573	12.9263	3.1142	4.1507	0.2425
1.35	0.7618	1.9596	1.2226	1.6028	0.9697	3.40	0.4552	13.3200	3.1802	4.1884	0.2322
1.40	0.7397	2.1200	1.2547	1.6897	0.9582	3.45	0.4531	13.7196	3.2472	4.2251	0.2224
1.45	0.7196	2.2863	1.2872	1.7761	0.9448	3.50	0.4512	14.1250	3.3151	4.2609	0.2129
1.50	0.7011	2.4583	1.3202	1.8621	0.9298	3.55	0.4492	14.5363	3.3839	4.2957	0.2039
1.55	0.6841	2.6363	1.3538	1.9473	0.9132	3.60	0.4474	14.9533	3.4537	4.3296	0.1953
1.60	0.6684	2.8200	1.3880	2.0317	0.8952	3.65	0.4456	15.3763	3.5245	4.3627	0.1871
1.65	0.6540	3.0096	1.4228	2.1152	0.8760	3.70	0.4439	15.8050	3.5962	4.3949	0.1792
1.70	0.6405	3.2050	1.4583	2.1977	0.8557	3.75	0.4423	16.2396	3.6689	4.4262	0.1717
1.75	0.6281	3.4063	1.4946	2.2791	0.8346	3.80	0.4407	16.6800	3.7426	4.4568	0.1645
1.80	0.6165	3.6133	1.5316	2.3592	0.8127	3.85	0.4392	17.1263	3.8172	4.4866	0.1576
1.85	0.6057	3.8263	1.5693	2.4381	0.7902	3.90	0.4377	17.5783	3.8928	4.5156	0.1510
1.90	0.5956	4.0450	1.6079	2.5157	0.7674	3.95	0.4363	18.0363	3.9694	4.5439	0.1448
1.95	0.5862	4.2696	1.6473	2.5919	0.7442	4.00	0.4350	18.5000	4.0469	4.5714	0.1388
2.00	0.5774	4.5000	1.6875	2.6667	0.7209	4.10	0.4324	19.4450	4.2048	4.6245	0.1276
2.05	0.5691	4.7363	1.7285	2.7400	0.6975	4.20	0.4299	20.4133	4.3666	4.6749	0.1173
2.10	0.5613	4.9783	1.7705	2.8119	0.6742	4.30	0.4277	21.4050	4.5322	4.7229	0.1080
2.15	0.5540	5.2263	1.8132	2.8823	0.6511	4.40	0.4255	22.4200	4.7017	4.7685	0.0995
2.20	0.5471	5.4800	1.8569	2.9512	0.6281	4.50	0.4236	23.4583	4.8751	4.8119	0.0917
2.25	0.5406	5.7396	1.9014	3.0186	0.6055	4.60	0.4217	24.5200	5.0523	4.8532	0.0846
2.30	0.5344	6.0050	1.9468	3.0845	0.5833	4.70	0.4199	25.6050	5.2334	4.8926	0.0781
2.35	0.5286	6.2763	1.9931	3.1490	0.5615	4.80	0.4183	26.7133	5.4184	4.9301	0.0721
2.40	0.5231	6.5533	2.0403	3.2119	0.5401	4.90	0.4167	27.8450	5.6073	4.9659	0.0667
2.45	0.5179	6.8363	2.0885	3.2733	0.5193	5.00	0.4152	29.0000	5.8000	5.0000	0.0617
2.50	0.5130	7.1250	2.1375	3.3333	0.4990	5.50	0.4090	35.1250	6.8218	5.1489	0.0424
2.55	0.5083	7.4196	2.1875	3.3919	0.4793	6.00	0.4042	41.8333	7.9406	5.2683	0.0297
2.60	0.5039	7.7200	2.2383	3.4490	0.4601	6.50	0.4004	49.1250	9.1564	5.3651	0.0211
2.65	0.4996	8.0262	2.2902	3.5047	0.4416	7.00	0.3974	57.0000	10.4694	5.4444	0.0154
2.70	0.4956	8.3383	2.3429	3.5590	0.4236	7.50	0.3949	65.4583	11.8795	5.5102	0.0113
2.75	0.4918	8.6562	2.3966	3.6119	0.4062	8.00	0.3929	74.5000	13.3867	5.5652	0.0085
2.80	0.4882	8.9800	2.4512	3.6636	0.3895	8.50	0.3912	84.1250	14.9911	5.6117	0.0064
2.85	0.4847	9.3096	2.5067	3.7139	0.3733	9.00	0.3898	94.3333	16.6927	5.6512	0.0050
2.90	0.4814	9.6450	2.5632	3.7629	0.3577	9.50	0.3886	105.1250	18.4915	5.6850	0.0039
2.95	0.4782	9.9862	2.6206	3.8106	0.3428	10.00	0.3876	116.5000	20.3875	5.7143	0.0030
3.00	0.4752	10.3333	2.6790	3.8571	0.3283	∞	0.3780	∞	∞	6.0000	0.0000